toxins can precipate intoxication or iatric disease must

d in an acute deliration were personal examination, many of the findings extening abnormal pulse oximetry and actory evaluation for normality was personal. Because cancer, a computed a performed. After story, it was apparis medical regiment the dosage. All of his stable dosage for 2 sumptive diagnosis

is medical regimen ch dosage. All of his stable dosage for 2 sumptive diagnosis to the transdermal yl metabolites was nd the patient was sive care unit. His removal of the fenions were continued d discharge at their s it unlikely that they to the delirium. An d-release morphine eliminated the pos moval of the fentany thout recurrence of

ission. onstrate evidence of inction of insufficient nselves, but possibly netabolism. Steinberg entanyl to norfentanyl on of norfentanyl, like tral nervous system of norfentanyl remain ted data qualifying of system effects of nor effect. Both fentany dine derivatives; their their metabolites are demonstrated to cause n stimulatory effects.† is rapidly metabolized to despropionyl norfentanyl, but this process may be impaired in patients with significant hepatic dysfunction.⁶ A literature review located a single brief case report of fentanyl-induced delirium caused by intravenous fentanyl.⁷ Although it is tempting to associate our patient's symptoms with a derangement in the metabolism or elimination of fentanyl or its metabolites, there is not sufficient information available to reach any valid conclusions, and to do so would be conjecture.

Clinicians who prescribe transdermal fentanyl should be aware of this potential complication. If central nervous system excitatory symptoms develop in a patient being treated with transdermal fentanyl, after other causes of delirium have been ruled out, consideration should be given to removing the patch and observing the patient while maintaining appropriate monitoring. An equianalgesic regimen of another opioid should be started to provide analgesia and prevent withdrawal.

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Intraabdominal Fire during Laparoscopic Cholecystectomy

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Received from Walter Reed Army Medical Center, Washington, D.C. Submitted for publication May 22, 1995. Accepted for publication May 27, 1995. The opinions expressed herein are the private views of the authors and are not to be construed as official or reflecting the views of the Department of the Army or the Department of Defense.

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Key words: Complications: carbon dioxide tanks; fire; laparoscopic surgery.

LAPAROSCOPIC surgery for elective cholecystectomy has undergone rapid acceptance since it initially was described in 1988.^{1–3} Complications unique to laparoscopic surgery result mainly from the cardiopulmonary impact of creating a pneumoperitoneum and choice of insufflating gas.^{4–6} Overall morbidity rates of 2–5% for laparoscopic cholecystectomy compare favorably to the 4–8% documented with open procedures.^{7–9} We report a case of intraabdominal fire secondary to insufflating the incorrect concentration of carbon dioxide gas during a laparoscopic cholecystectomy.

This article is accompanied by a letter to the editor. Please see: Johnson CT: Reply from Compressed Gas Association, Inc. Anesthesiology 83:879, 1995.

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Case Report

This case involves a 62-yr-old, 114-kg, 150-cm man who presented with resolving acute pancreatitis and gallstones for elective laparoscopic cholecystectomy. Physical examination was significant for obesity and dullness to percussion at both lung bases. After induction of anesthesia and tracheal intubation, anesthesia and neuromuscular blockade were maintained with 0.5-1% isoflurane in 40% O_2 in nitrogen and vecuronium as needed to maintain two twitches in response to train-of-four stimulation.

Pneumoperitoneum was achieved with a Storz Electronic Laparoflator (Karl Storz Endoscopy-America, Culver City, CA), which was connected to the carbon dioxide tank by a collar specifically pin-indexed for carbon dioxide. The necessary laparoscopic equipment was maintained on a metal cart (Medi-Mech, The Dalles, OR). The carbon dioxide cylinder was located behind the laparoscopic cart, in a metal jacketed well that concealed the lower 90% of the tank. Approximately 2 h into what had been an uneventful surgery, a 2-cm "flame" arising from the tip of the electrocautery unit was visualized on the video monitor. Seconds later, as the electrocautery was being rapidly withdrawn through the trocar, the entire video screen turned orange-red and the abdominal wall was transilluminated. The presumed fire lasted approximately 2 s. Direct visual inspection of the electrocautery tip and trocar (fig. 1) revealed that both of these instruments had undergone combustion. The abdomen was immediately

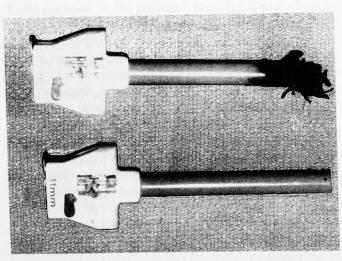


Fig. 1. Charred upper trocar, which was removed immediately from the peritoneal cavity after transillumination of the abdomen. An uncharred lower trocar is displayed for comparison.

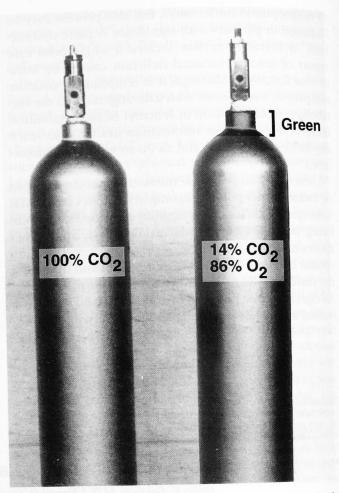


Fig. 2. Comparison of 100% CO₂ tank (*left*) and 14% CO₂ tank (*right*). Note green band at the top of the cylinder on the right, denoting the presence of oxygen. Pin-index systems for both tanks are identical.

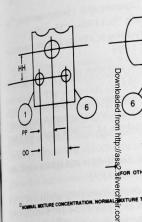
opened through a right subcostal incision. Visual inspection of the abdomen revealed no evidence of thermal injury with the exception of minor charring of the falciform ligament. Gross examination of liver, stomach, pancreas, and intestines was unremarkable. The gallbladder was removed, the peritoneal cavity lavaged with 3 liters of normal saline irrigation, and the abdomen closed in standard two-layer fashion. The remaining intraoperative and postanesthesia course proceeded without incident. The patient was discharged from the hospital on the 9th postoperative day without further complication.

After the incident, the carbon dioxide cylinder was removed from its protective metal jacket mounted on the back of the insufflator cart. Close inspection revealed a 3 cm-wide green stripe at the top of the tank, indicating the presence of oxygen (fig. 2). The re-

CASE REPORTS

PIN-INI STANDARD MEDICAL PRESSURES

Carbon Die



mainder of the cylinder was contained carbon diogide. cealed label on the carbon the gas was a mixture of 14% phone call to the gas suppany cylinder with a carbon than 7% is fitted with standioxide (fig. 3). Comparison tained 14% CO₂, with anote (fig. 2) confirmed that the property of the potential for this typ

Discussion

Several reports in the 1 complications during laparathe result of either paeum flating gas. 10-14 The first do dominal explosion was a no in 1933 when Ferver used gas. 10 The use of carbon d with both fatal and nonfatal aparoscopic surgery. 11.14 Microsophic flating in the 1970s. 11-13 Current 100% CO₂, because it is no not create an explosion sho

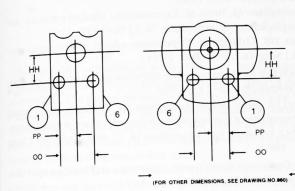
§ Manual of the Compressed G pressed Gas Cylinder Valve Outle ton, D.C., 1994.

Anesthesiology, V 83, No 4, Oct

PIN-INDEXED YOKE, PINS 1-6

STANDARD MEDICAL CYLINDER VALVE YOKE CONNECTION FOR PRESSURES UP TO 3,000 psig (20 680 kPa) FOR Carbon Dioxide

Carbon Dioxide & Oxygen Mixtures (C02 over 7%°)



DIMEN- SIONS	INCHES	ММ
нн	.487 + .033	12.4 + .07
PP	.281 + .005	7.15 + .15
00	.562 + .003	14.3 + .07

Fig. 3. Diagram of the Compressed Gas Association (CGA) specifications for tanks containing >7% CO₂ (CGA 940). The diagram on the right specifies the dimensions of the inlet valve located on the carbon dioxide collar. The diagram on the left specifies the dimensions of the outlet valve located on the carbon dioxide cylinder yolk.

ONOMINAL MIXTURE CONCENTRATION, NORMAL MIXTURE TOLERANCES ARE ALLOWABLE

mainder of the cylinder was gray, indicating the tank contained carbon dioxide. Examination of the concealed label on the carbon dioxide tank revealed that the gas was a mixture of 14% CO $_2$ and 86% O $_2$. A telephone call to the gas supply company revealed that any cylinder with a carbon dioxide content of greater than 7% is fitted with a standard pin-index for carbon dioxide (fig. 3). Comparison of our tank, which contained 14% CO $_2$, with another containing 100% CO $_2$ (fig. 2) confirmed that the pin-indexes were identical. None of the involved operating room staff were aware of the potential for this type of error.

Discussion

Several reports in the literature document major complications during laparoscopic surgery that were the result of either pneumoperitoneum or the insufflating gas. ^{10–14} The first documented case of intraabdominal explosion was a nonfatal event that occurred in 1933 when Ferver used 100% O₂ as the insufflating gas. ¹⁰ The use of carbon dioxide has been associated with both fatal and nonfatal injuries sustained during laparoscopic surgery. ^{11,14} Nitrous oxide has been implicated in several intraabdominal explosions described in the 1970s. ^{11–13} Current surgical practice is to use 100% CO₂, because it is noncombustible and thus will not create an explosion should the electrocautery gen-

erate a spark or ignite bowel gas (methane/hydrogen).¹⁵ Diagnostic endoscopists prefer nitrous oxide, because it is better tolerated (less peritoneal irritation) in their awake patients and because they do not use electrocautery.

Several industry standards have been written and adopted to prevent the use of gases other than 100% CO2 for laparoscopic surgery.§ In addition, standard operating procedure at our institution only permits the placement of 100% CO2 cylinders in our operating theater. Gas cylinders with a mixture of carbon dioxide and oxygen come from the distributor with a small green stripe at the top (fig. 2), indicating the presence of oxygen. The distributor also provides a large centrally placed label on the tank identifying the specific gas mixture within. As a further precaution, the gas insufflator collar uses the Compressed Gas Association (CGA) pin-index system, which is specific for carbon dioxide (CGA 940) concentrations greater than 7% (fig. 3). Operating room support personnel (nurses and technicians) are responsible for transporting the carbon dioxide cylinders from the storage area, mounting them on the laparoscopic carts and connecting them to the insufflating device.

Insufflation of this patient's peritoneal cavity with a gas mixture containing less than 100% CO₂ illustrates that the current safety features are not foolproof. Human error permitted a tank with a carbon dioxide/oxygen mixture into our operating theater. A member of the operating room support personnel mistakenly placed the incorrect cylinder into a metal jacket, concealing the tank's centrally placed label. This left only the thin green stripe at the top of the tank to warn

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Green

Green

Green

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I incision. Visual in-I no evidence of therminor charring of the nation of liver, stoms unremarkable. The itoneal cavity lavaged rigation, and the abayer fashion. The renesthesia course pronetient was discharged operative day without

dioxide cylinder was all jacket mounted on Close inspection for at the top of the tank, agen (fig. 2). The fig.

[§] Manual of the Compressed Gas Association: Standards for Compressed Gas Cylinder Valve Outlet and Inlet Connections. Washington, D.C., 1994.

members of the operating room staff of the presence of oxygen. Finally, use of the current medical gas pinindex system, which does not distinguish tanks containing 100% CO₂ from those containing carbon dioxide mixtures between 7% and 99% permitted us to connect a tank that contained 86% O₂ to the insufflation device.

The presence of a pin-index configuration specific for 100% CO2 would have prevented our case of intraabdominal fire. The absence of documentation in the literature of similar disasters infers that the human error made in our case represents a rare breach in existing preventive measures. However, as the number of laparoscopic procedures approaches 1,000,000 per year, the likelihood of an event such as the one we report here will increase. Until a 100% CO₂-specific pin-index fitting or appropriate safety mechanism is designed, avoidance of an intraabdominal fire relies solely on the vigilance of informed personnel. Review of this case by our quality improvement program resulted in inservice training of all operating room personnel (physicians, nurses, technicians) to alert them of this potential hazard. In addition, a more stringent check-in procedure for gas cylinders used in the operating theater was established. We report this case of intraabdominal fire during laparoscopic cholecystectomy to alert the anesthesia community of its potential to occur and to prompt efforts to eliminate this rare but potentially devastating complication.

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CORRESPONDE

Mosthesiology 8§875, 1995 6 1995 American Society of Anesthesio Uppincott-Raven Publishers

Propofol in

To the Editor: - Poisoning wit is well known to produce coma w convulsions. The prognosis usual tomatic treatment is instituted earl with success, to control mgoclon poisoning in which diazepam alo ceeded in controlling myor A 70-yr-old woman with a hist admitted to our intensive care un Four hours after ingestion of chlo was discovered at home, uniconse contractions. After 20 mg∃intrav intubated and the lungs were med роп, despite a diazepam affusio peared. At the hospital, gas Fic lav diazepam (40-mg bolus) and co admission to the intensived care i diazepam over 3 h of 146 mg (occurred and were increased by diazepam was stopped, and prope patient received a loading sose o within 1 min. no new myoelonic stimuli. After 10 min, hosever, observed again. A second dese of j by an infusion of 1 mg·kg-1·h further myoclonic jerks oceurred continued, after which awakening extubated 6 h after admission, an the intensive care unit after 25 h Propofol has been noted to have its use in the management of refi reported.3 In this case, propofol c by chloralose poisoning after a l

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To the Editor:—Hynson et al. anism of the aortoradial pressure at the conclusion of cardiopulmo femoral and radial arterial pressure blood flows in six healthy 20–40 cold temperature (21°C) until shi

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